

S. S. College. Jehanabad (Magadh University)

Department : Physics

Subject : Thermodynamics

Class : B.Sc(H) Physics Part I

Topic: Application of Maxwell's Thermodynamical Relation

Teacher : M. K. Singh

- Using Maxwell's thermodynamical relations, To show that

$$\left(\frac{\partial C_v}{\partial V}\right) = T \left(\frac{\partial^2 S}{\partial V \partial T}\right) = T \left(\frac{\partial^2 P}{\partial T^2}\right)_v$$

From Maxwell's relation we have

$$\left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial P}{\partial T}\right)_v \dots\dots\dots (i)$$

Differentiating equation (i) with respect to temperature we have

$$\left(\frac{\partial^2 S}{\partial T \partial V}\right) = \left(\frac{\partial^2 P}{\partial T^2}\right)_v \dots\dots\dots (ii)$$

We know that

$$C_v = T \left(\frac{\partial S}{\partial T}\right)_v \dots\dots\dots (iii)$$

Differentiating equation (iii) with respect to volume we have

$$\left(\frac{\partial C_v}{\partial V}\right) = T \left(\frac{\partial^2 S}{\partial T \partial V}\right) \dots\dots\dots (iv)$$

From equation (ii) and (iv) we have

$$\begin{aligned} \left(\frac{\partial C_v}{\partial V}\right) &= T \left(\frac{\partial^2 S}{\partial T \partial V}\right) \\ &= T \left(\frac{\partial^2 P}{\partial T^2}\right)_v \dots\dots\dots (v) \end{aligned}$$

- Using Maxwell's thermodynamical relations, To show that

$$\left(\frac{\partial C_P}{\partial P}\right) = \left(\frac{\partial^2 S}{\partial P \partial T}\right) = -T \left(\frac{\partial^2 V}{\partial T^2}\right)_P$$

Sol. From Maxwell's relations we have

$$\left(\frac{\partial S}{\partial P}\right)_T = -\left(\frac{\partial V}{\partial T}\right)_P \dots\dots\dots (i)$$

Differentiating equation (i) with respect to temperature we get

$$\left(\frac{\partial^2 S}{\partial T \partial P}\right) = -\left(\frac{\partial^2 V}{\partial T^2}\right)_P \dots\dots\dots (ii)$$

We know that

$$C_P = T \left(\frac{\partial S}{\partial T}\right)_P \dots\dots\dots (iii)$$

Differentiating equation (iii) with respect to pressure we get

$$\left(\frac{\partial C_P}{\partial P}\right) = T \left(\frac{\partial^2 S}{\partial T \partial P}\right) \dots\dots\dots (iv)$$

From equations (ii) and (iv) we get

$$\begin{aligned} \left(\frac{\partial C_P}{\partial P}\right) &= T \left(\frac{\partial^2 S}{\partial T \partial P}\right) \dots\dots\dots (v) \\ &= T \left(\frac{\partial^2 V}{\partial T^2}\right)_P \end{aligned}$$